Tracer relationships in surface waters of coastal waters from the Gulf of Alaska, Bering and Chukchi Seas

Burke Hales¹

Laurie Juranek¹

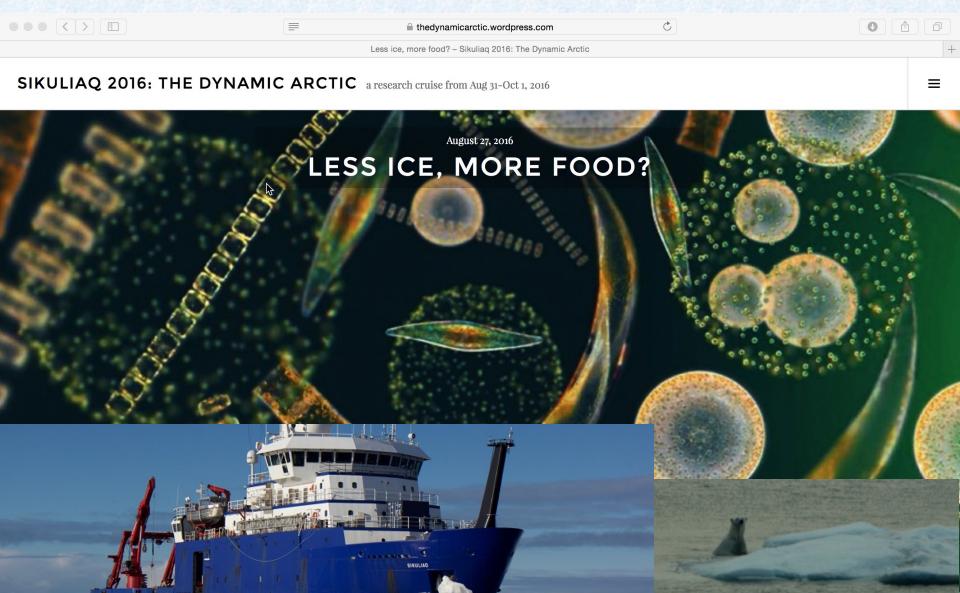
Selina Lambert¹, Carrie Weekes^{1,2}, and Katie Pocock²

- 1: Oregon State University College of Oceanic and Atmospheric Sciences
- 2: Hakai Institute of the Tula Foundation



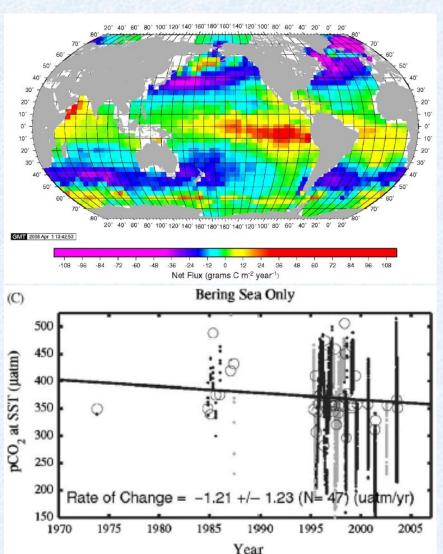
A Stirred Up Arctic

Does reduced ice cover in late-summer coastal Arctic waters allow a second burst of productivity as katabatic winds mix exposed waters and re-supply nutrients to the surface?



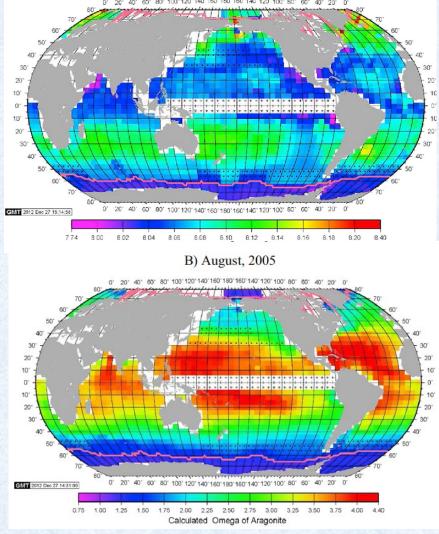
Relevance of high-latitude carbonate chemistry?

Air-sea CO₂ exchange. High-lat areally-disproportionate sink, may be intensifying



Takahashi et al., 2009

Ocean acidification. Arctic has seemingly benign pH, but low shell-mineral stability



Takahashi et al., 2014

New monitoring approaches

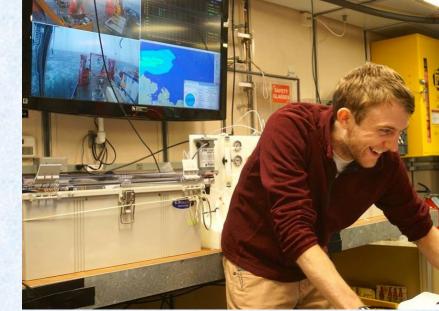
Full constraint of carbonate system requires determination of two independent parameters; 'safest' two are CO_2 partial pressure (pCO_2) and total dissolved carbonic acid (TCO_2)



Combined pCO₂/TCO₂ measurement on continuous flow and/or discrete samples (aka 'Burke-o-Lator').

State of the art accuracy and precision (0.2% in TCO_2 , and < 1% in pCO_2).

Two parallel systems aboard RV Sikuliaq, one in continuous TCO₂ mode, one in continuous pCO₂



Cruise description



Two cruises, Nome-Nome:

3-24 September 2016

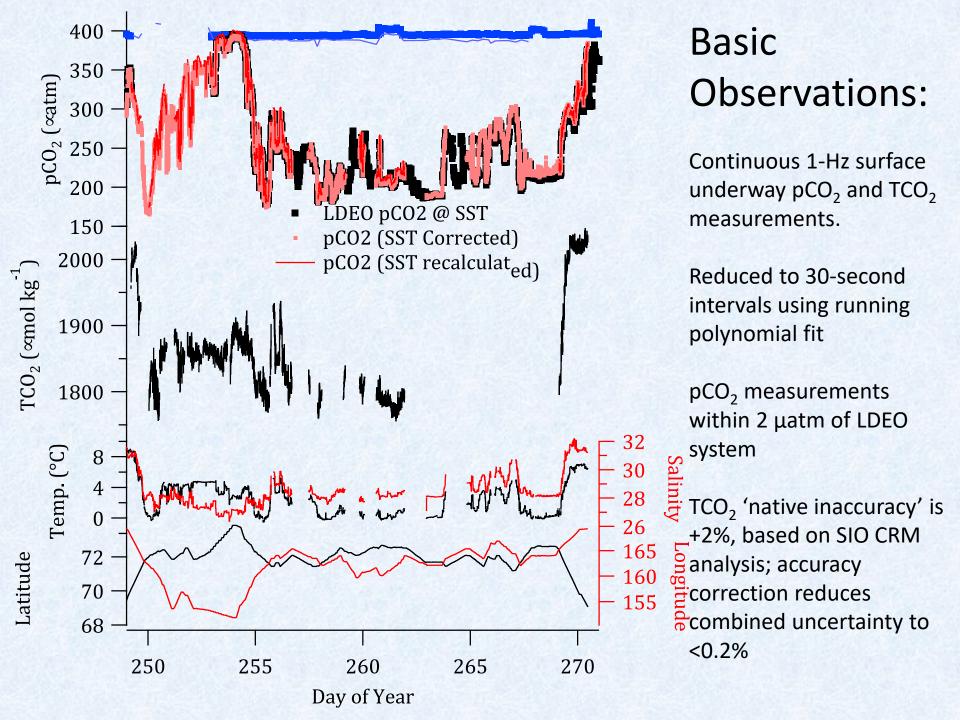
-Focus on Chukchi Sea waters > 20 nm offshore, but with excursion to Beaufort Sea and Arctic Basin.

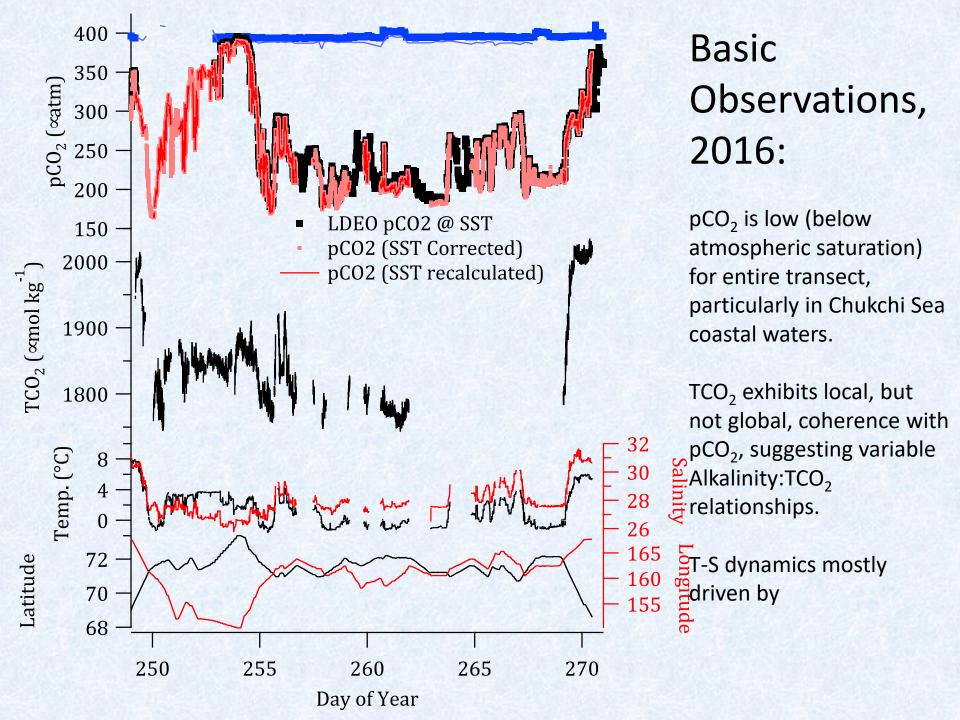
-Significant wind-transport of ice into study area

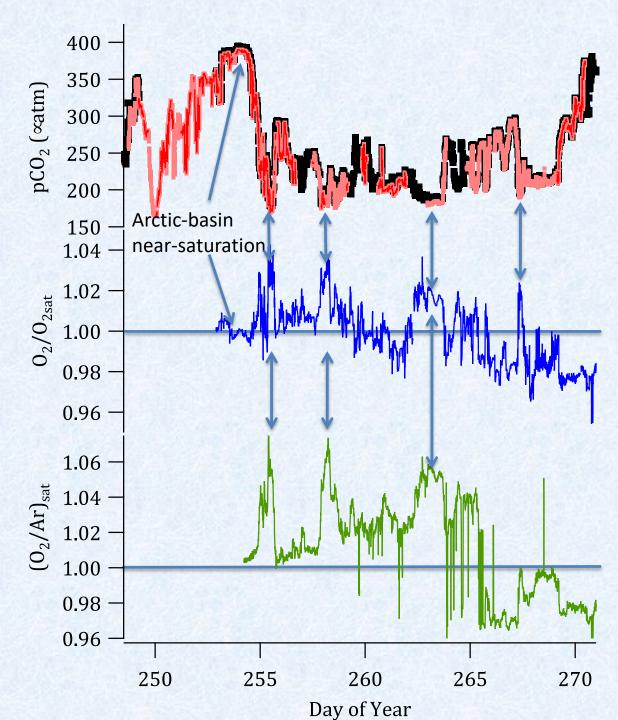
8-22 August 2017

Focus on Chukchi Sea waters closer to shore.

No sea ice in study area.







Biological and physical drivers

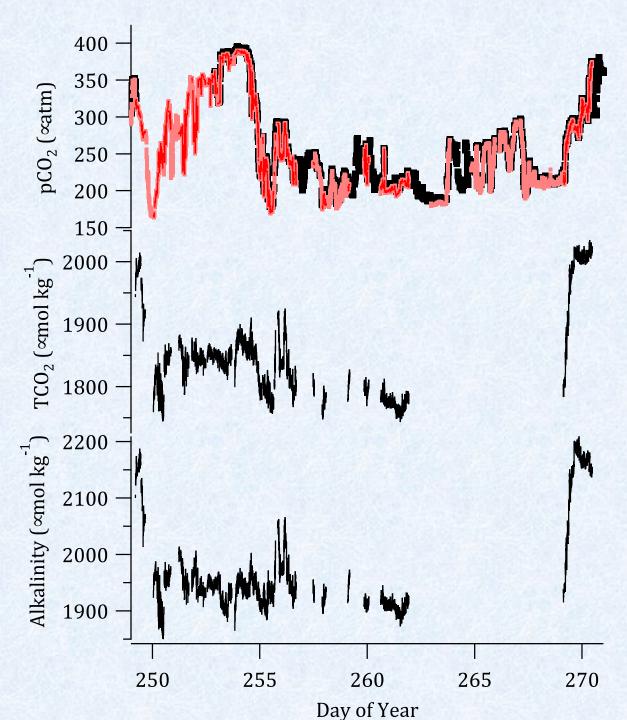
Low pCO₂ is the result of remnant net productivity and slow gas-exchange time-scales for CO₂.

Local features show strong coherence.

Normalization of O₂ to the inert Ar highlights the role of biological processes

Differential gas exchange timescale explains the loss of O_2 supersaturation while pCO_2 remains low.

Late O₂/Ar undersaturation implies development of respiration signal

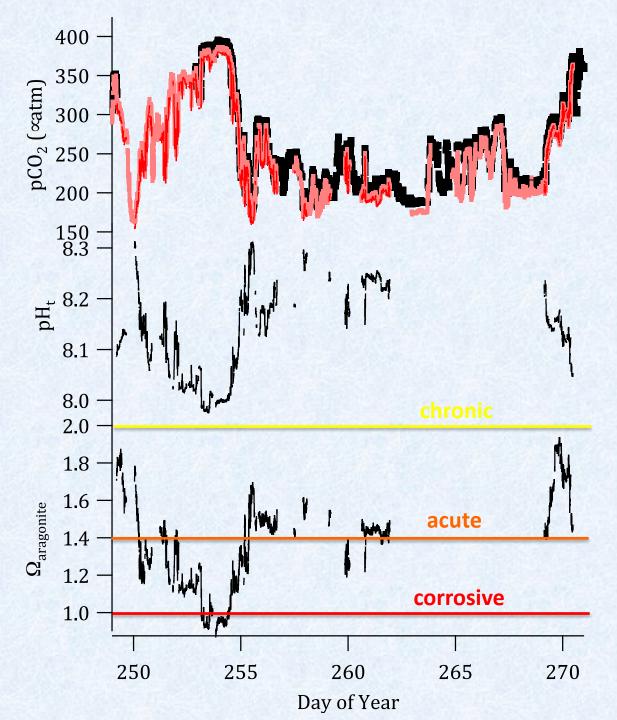


Calculated parameters

With coincident TCO₂ and pCO₂ (and T and S) we can recalculate other parameters of the carbonate system.

Alkalinity is the most basic of these parameters, as it is Temperature-independent, and displays quasi-conservative behavior with respect to Salinity.

Its determination allows recalculation of pCO₂ @ in situ temperatures, and local determination of the T-correction coefficient.



Calculated parameters

Other parameters of interest can be calculated as well.

The overall low pCO₂ corresponds with an overall high pH. Good news for OA, right?

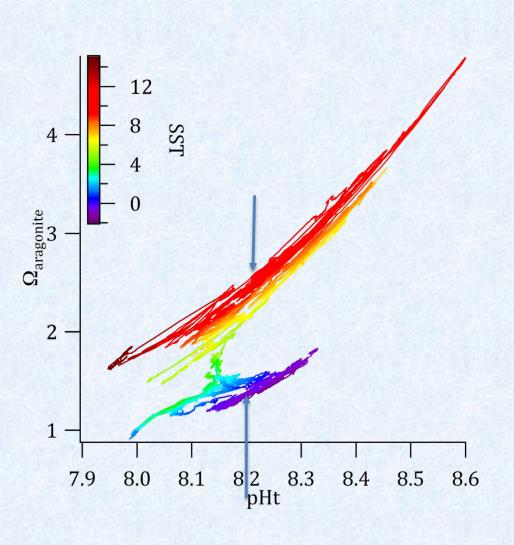
Wrong. The most proximally important parameter, shell-mineral stability (aka Ω), is low.

Always below chronic-effect threshold in 2016.

~50% of time below acute threshold

Sometimes even corrosive!

Failure of single-proxy assessment of Ocean Acidification status



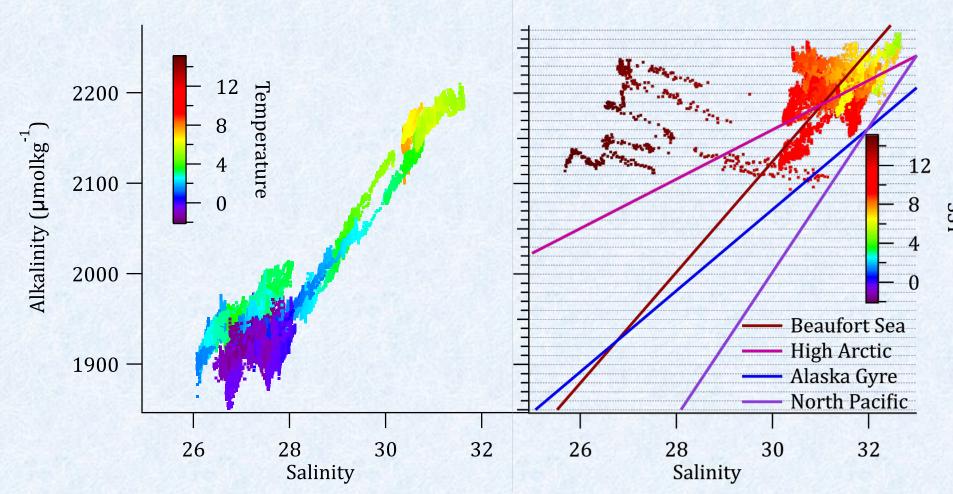
One last reminder of the need for full CO₂-system constraint.

Different Ω for same pH among different water masses (>100% difference in Ω for common pH).

Different Ω -pH relationships across water masses, between years.

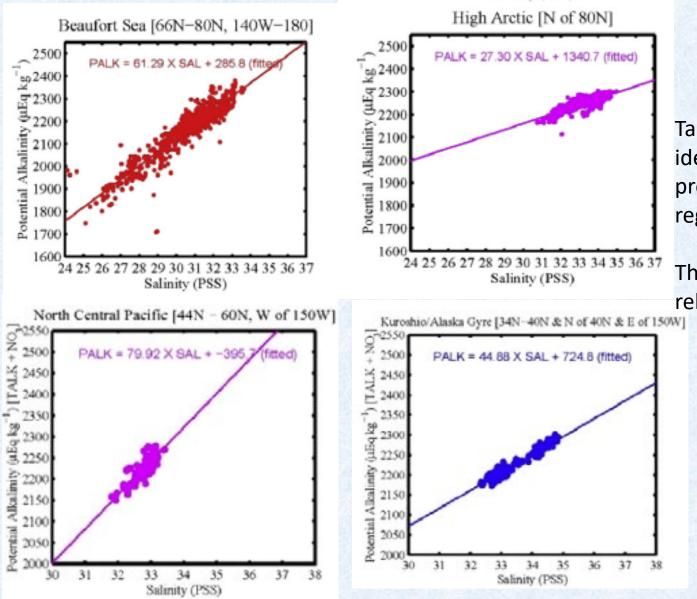
Alkalinity - Salinity

Alkalinity is semi-conservative, and can be estimated from empirical dependences on salinity



Recent results show several distinct Alk-S characteristics

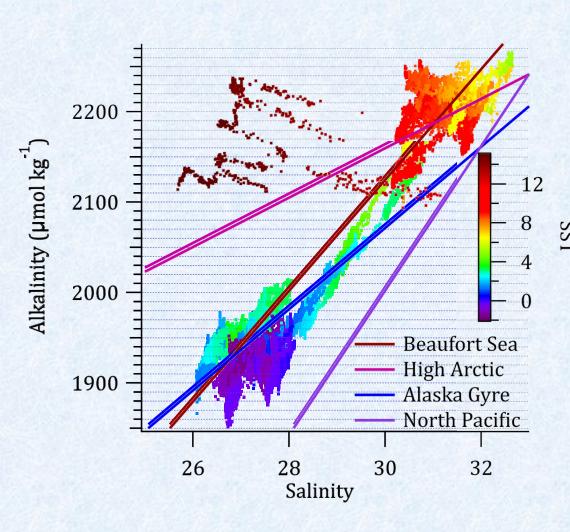
Sources of variability in the alk-S relationship



Takahashi et al. 2014 identified several proximally-relevant subregimes.

These have distinct Alk-S relationships

Our data within this context



Best overall agreement Takahashi's historical data is with the 'Beaufort Sea'

Significant areas of departure from published relationships, mostly negative in 2016: As much as 80 μ mol kg⁻¹ (0.15 error in Ω)

Positive deviations in 2017, both at high and low-S

What are these? Calcification? De-nitrification? Multiple end-member mixing? High-Alk terrestrial freshwaters?

Conclusions

- First-ever high-resolution continuous combined pCO₂ and TCO₂ measurements in Arctic surface waters show low pCO₂, persistently below atmospheric saturation.
- pCO₂ is low because of recent net community photosynthesis; persists because of long gas-exchange times for CO₂.
- Combination of pCO₂ and TCO₂ allows most robust calculation of other carbonate-system parameters like alkalinity, pH, and Ω .
- Low pCO_2 and high pH belie low Ω , frequently below thresholds identified for harmful impacts on planktonic larval bivalves, and illustrating the need for full carbonate-system constraint.
- Alk-S relationships show significant departures from historical data; causes uncertain.

Acknowledgements

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